

## 55. Photochemistry of 2,2-Dimethyl-2*H*-furo[3,4-*b*]pyran-4,7(3*H*,5*H*)-dione

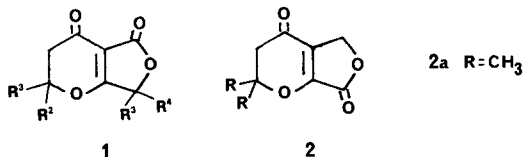
by Sabine Lau and Paul Margaretha\*

Institut für Organische Chemie, Universität, D-2000 Hamburg 13

(15.XII.87)

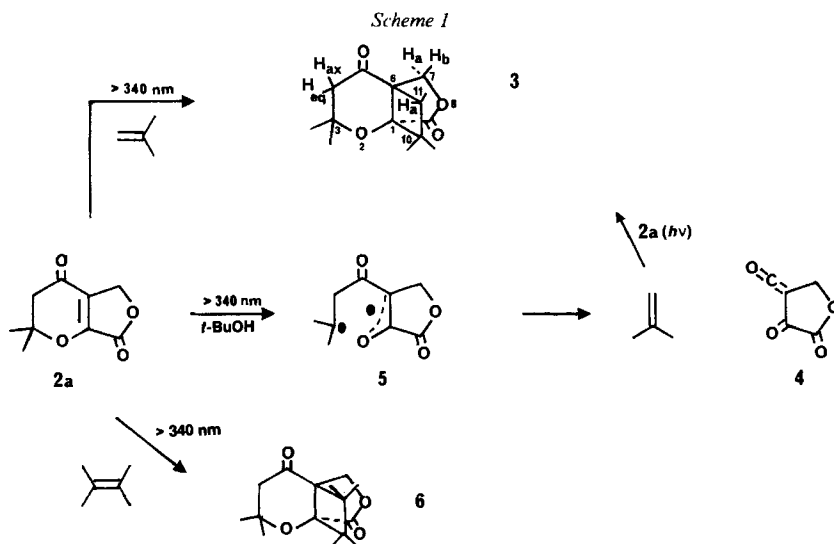
Irradiation ( $\lambda > 340$  nm) of the title compound **2a** in *t*-BuOH affords 3,3,10,10-tetramethyl-2,8-dioxatricyclo-[4.3.2.0<sup>1,6</sup>]undecane-5,9-dione (**3**) via photoextrusion of 2-methylpropene and subsequent photocycloaddition of **2a** to the alkene. The same regioisomer **3** is formed selectively and in much higher yields on irradiating **2a** in the presence of excess 2-methylpropene. Irradiation of **3** ( $\lambda = 300$  nm) in *t*-BuOH in the presence of the same alkene gives a 6:1 mixture of spirooxetanes **7a** and **7b** but not  $\alpha$ -cleavage products. In 2-propanol, **2a** is photoreduced to a 2:1 mixture of diastereoisomeric hydrodimers **9**.

For some time, we have been interested in 2,3-dihydro-4-pyranones [1–4] and in 2(5*H*)-furanones [5] as model compounds in [2 + 2] photocycloaddition reactions. Two type of molecules, both isomers of the antibiotic patulin [6], contain these two ring systems fused *via* the common C–C bond, the 2*H*-furo[3,4-*b*]pyran-4,5(3*H*,7*H*)-diones **1** [7] and the corresponding furopyran-4,7(3*H*,5*H*)-diones **2** [8] [9]. We now report on the photochemistry of the title compound **2a**.



Irradiation ( $\lambda > 340$  nm) of **2a** in *t*-BuOH leads to decomposition of the starting material with formation of one new product **3**, albeit in low yield (15%). The CI-MS of **3** shows [ $M + H^+$ ] at  $m/z$  239 and the <sup>1</sup>H-NMR indicates the presence of 1 additional CH<sub>2</sub> group and 2 additional CH<sub>3</sub> groups. Both results combined suggest that **3** is formed from **2a** and 2-methylpropene. Indeed, irradiation of **2a** in *t*-BuOH saturated with this alkene affords **3** much faster and in very good yield. Competitive experiments in a 'merry-go-round' setup show that for total conversion of **2a** to **3** in the presence of alkene, only 10% of **2a** are decomposed in its absence with very little formation of **3**. It, thus, becomes evident that **2a** undergoes photoextrusion of 2-methylpropene and that remaining **2a** undergoes [2 + 2] photocycloaddition to this alkene. The primary reaction product **4** apparently decomposes quantitatively to CO, CO<sub>2</sub> and CH<sub>2</sub>=C=C=O (*Scheme 1*).

The first step in the photoextrusion of 2-methylpropene from **2a** is the cleavage of the (CH<sub>3</sub>)<sub>2</sub>C–O bond to give the alkyl-vinyloxy biradical **5**. This is in analogy to the behaviour of benzyl enol ethers which undergo 1,3-migration of the benzyl group on irradiation

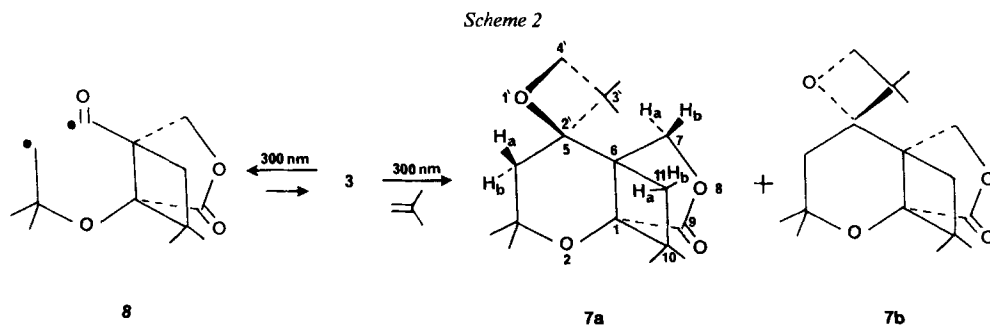


[10]. For biradical **5**,  $\beta$ -cleavage to 2-methylpropene and **4** seems to be the preferred path over (a possible) spirocyclobutanone formation.

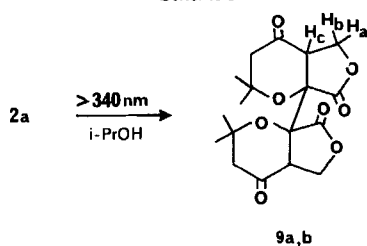
Cycloaddition of **2a** to 2-methylpropene is regioselective, the orientation of addition being the one expected for the excited pyranone – but not for the furanone moiety [4]. The structure of **3** is proven by its  $^1\text{H-NMR}$  spectrum wherein a *W*-coupling between  $\text{H}_a\text{C}(7)$  and  $\text{H}_a\text{C}(11)$  with  $J(1,5) = 1.0$  is observed. No such coupling is observed for cycloadduct **6** formed from **2a** and 2,3-dimethyl-2-butene.

When the irradiation of **2a** and 2-methylpropene is performed with light of  $\lambda = 300 \text{ nm}$ , **3** is again formed selectively but then undergoes a consecutive photocycloaddition to the alkene to afford a 6:1 mixture of the diastereoisomeric spirooxetanes **7a** and **7b** (Scheme 2). The relative configuration of the major product **7a** again results from  $^1\text{H-NMR}$ , wherein  $\text{H}_a\text{C}(11)$  and  $\text{H}_a\text{C}(4)$  are deshielded due to the oxetane O-atom by 1.2 and 1.0 ppm as compared to  $\text{H}_b\text{C}(11)$  and  $\text{H}_b\text{C}(4)$ , respectively. The MS of **7a** and **7b** are alike.

It is interesting to note that no  $\alpha$ -cleavage products are formed from **3**. This reaction occurs efficiently with strained bicyclo[4.2.0]octan-2-ones in the 'abnormal' mode



Scheme 3

Table. Spectroscopic Data of Compounds **2a**, **3**, **6**, **7**, and **9**

Compound <sup>a)</sup>	UV (CH <sub>2</sub> Cl <sub>2</sub> )	IR (KBr)	MS	<sup>1</sup> H-NMR (CDCl <sub>3</sub> )	<sup>13</sup> C-NMR (CDCl <sub>3</sub> )
<b>2a</b>	325 (34), 290 (sh, 4900), 280 (6600), 274 (sh, 6000)	1760, 1691, 1649	182 (40, M <sup>+</sup> ), 56	4.99 (s, 2 H); 2.67 (s, 2 H); 1.57 (s, 6 H)	190, 168, 155, 124, 88, 67 (t), 49 (t), 26 (q)
<b>3</b>		1775, 1700	238 (1, M <sup>+</sup> ), 183	4.78 (dd, J = 10.6, 1.0); 4.46 (d, J = 10.6); 2.76 (dq, J = 15.6, 0.9); 2.37 (d, J = 15.6); 2.32 (dd, J = 12.5, 1.0); 2.02 (d, J = 12.5); 1.49, 1.28, 1.22 (3s, CH <sub>3</sub> ); 1.08 (d, J = 0.9, CH <sub>3</sub> )	206, 175, 81, 79, 73 (t), 49 (t), 43, 42 (t), 38, 30 (q), 26 (q), 25 (q), 24 (q)
<b>6</b>		1765, 1700	266 (1, M <sup>+</sup> ), 183	4.70, 4.32 (AB, J = 11.0); 2.50 (dq, J = 16.8, 0.8); 2.42 (d, J = 16.8); 1.46, 1.18, 1.13, 1.12, 1.09 (5s, CH <sub>3</sub> ); 1.12 (d, J = 0.8, CH <sub>3</sub> )	
<b>7a</b>		1762 <sup>b)</sup>	no M <sup>+</sup> at 294, 239 (20), 183 (95), 56	4.30 (d, J = 10.4); 4.05 (dd, J = 10.5, 1.0); 4.30, 4.02 (AB, J = 5.6); 2.96 (dd, J = 12.0, 1.0); 2.37, 1.38 (AB, J = 14.8); 1.72 (d, J = 12.0); 1.48, 1.22, 1.15, 1.11, 1.10, 1.04 (6s, CH <sub>3</sub> )	
<b>9a</b>		1765, 1718 <sup>c)</sup>	367 (100, [M + 1] <sup>+</sup> <sup>d)</sup>	5.02 (dd, J = 8.7, 1.7); 4.44 (dd, J = 8.7, 8.2); 4.23 (dd, J = 8.2, 1.7); 2.49, 2.40 (AB, J = 14.0); 1.40, 1.07 (2s, CH <sub>3</sub> )	
<b>9b</b>				4.83 (dd, J = 9.5, 2.5); 4.47 (dd, J = 9.5, 9.1); 4.00 (dd, J = 9.1, 2.5); 2.64, 2.47 (AB, J = 13.9); 1.46, 1.07 (2s, CH <sub>3</sub> )	

<sup>a)</sup> All new compounds gave satisfactory elemental analyses.

<sup>b)</sup> In CH<sub>2</sub>Cl<sub>2</sub>.

<sup>c)</sup> Mixture **9a/9b**.

<sup>d)</sup> CI-MS spectrum.

[11–13]. Thus, biradical **8** should be formed, but this species cannot rearrange neither to a ketene nor to an unsaturated aldehyde and, therefore, will reclose to **3** (*Scheme 2*).

Finally, irradiation ( $\lambda > 340$  nm) of **2a** in 2-propanol gives a precipitate consisting of a 2:1 mixture **9a/9b**. Both compounds **9** have the molecular weight  $M^+ 366$ . The IR spectrum indicates both a ketonic and a lactone CO group and the  $^1\text{H-NMR}$ , recorded immediately after dissolving **9** in  $\text{CDCl}_3$  as decomposition occurs on standing at r.t. in  $\text{CDCl}_3$  or  $\text{CD}_3\text{COCD}_3$ , indicates an *ABC* pattern for the  $\text{CH}_2$  group next to the O-atom. These results suggest that **9a** and **9b** are *meso* and (*RS, RS*) hydrodimers (*Scheme 3*).

The spectroscopic data of compounds **2a**, **3**, **6**, **7**, and **9** is summarized in the *Table*.

### Experimental Part

*General.* See [5]. Anal. GC: 25-m *SE 30* capillary column.

*Starting Materials.* Furopyrandione **2a** was synthesized according to [9].

*Photolyses.* These were performed in a *Rayonet-RPR-100* photoreactor using either 350-nm lamps in combination with a liquid filter with cut-off at  $\lambda = 340$  nm [14] (irradiation of **2a**) or  $\lambda = 300$  nm lamps (irradiation of **3**).

*Irradiation of 2a.* An Ar-degassed soln. of 182 mg (1 mmol) of **2a** in 5 ml of *t*-BuOH is irradiated ( $\lambda > 340$  nm) for 80 h. Evaporation of the solvent and recrystallization of the residue from  $\text{Et}_2\text{O}$ /pentane affords 30 mg (13%) of 3,3,10,10-tetramethyl-2,8-dioxatricyclo[4.3.2.0<sup>1,6</sup>]undecane-5,9-dione (**3**). M.p. 128–130°.

*Irradiation of 2a and 2-Methylpropene.* A soln. of 910 mg (5 mmol) of **2a** in 10 ml of *t*-BuOH saturated with 2-methylpropene is irradiated ( $\lambda > 340$  nm) for 16 h. Work up as above affords 904 mg (76%) of **3**.

*Irradiation of 2a and 2,3-Dimethyl-2-butene.* An Ar-degassed soln. of 910 mg (5 mmol) of **2a** and 2.1 g (25 mmol) of 2,3-dimethyl-2-butene in 10 ml of *t*-BuOH is irradiated ( $\lambda > 340$  nm) for 32 h. Workup as above affords 905 mg (68%) of 3,3,10,10,11,11-hexamethyl-2,8-dioxatricyclo[4.3.2.0<sup>1,6</sup>]undecane-5,9-dione (**6**). M.p. 65–68°.

*Irradiation of 2a in 2-Propanol.* An Ar-degassed soln. of 910 mg (5 mmol) **2a** in 10 ml of 2-propanol is irradiated for 14 h. Filtration and drying of the white crystalline precipitate affords 732 mg (80%) of a 2:1 mixture (GC) of diastereoisomeric hydrodimers **9a/9b**. M.p. 185° (dec.). Compounds **9** decompose at r.t. in  $\text{CHCl}_3$  or acetone soln. to unknown products.

*Irradiation of 3 and 2-Methylpropene.* A soln. of 476 mg (2 mmol) of **3** in 5 ml of *t*-BuOH saturated with 2-methylpropene is irradiated ( $\lambda = 300$  nm) for 40 h to afford a 6:1 mixture (GC) **7a/7b**. Chromatography ( $\text{SiO}_2$ ,  $\text{CH}_2\text{Cl}_2$ ) affords 282 mg (48%) 3,3,3',3',10,10-hexamethylspiro[2,8-dioxatricyclo[4.3.2.0<sup>1,6</sup>]undecane-5,2'-oxetane]-9-one (**7a**). M.p. 81–84°. The minor component **7b** was not isolated.

### REFERENCES

- [1] P. Margaretha, *Tetrahedron* **1973**, *29*, 1317.
- [2] P. Margaretha, *Liebigs Ann. Chem.* **1973**, 727.
- [3] P. Margaretha, *Helv. Chim. Acta* **1974**, *57*, 2237.
- [4] P. Margaretha, *Chimia* **1975**, *29*, 203.
- [5] E. Anklam, P. Margaretha, *Helv. Chim. Acta* **1983**, *66*, 1466.
- [6] P. Coudert, J. Couquelet, J. Fialip, F. Sannajust, P. Bastide, A. Eschalièr, *Farmaco, Ed. Sci.* **1987**, *42*, 145.
- [7] S. Gelin, P. Pollet, *J. Heterocycl. Chem.* **1975**, *12*, 787.
- [8] B. Puetzer, C.H. Nield, R.H. Barry, *J. Am. Chem. Soc.* **1945**, *67*, 832.
- [9] J. Couquelet, P. Tronche, *Bull. Soc. Chim. Fr.* **1968**, 2015.
- [10] J.T. Pinhey, K. Schaffner, *Austr. J. Chem.* **1968**, *21*, 2265.
- [11] P.J. Nelson, D. Ostrem, J.D. Lassila, O.L. Chapman, *J. Org. Chem.* **1969**, *34*, 811.
- [12] R.J. Batten, H.A.J. Carless, *J. Chem. Soc., Chem. Commun.* **1985**, 1146.
- [13] G. Cruciani, C. Semisch, P. Margaretha, manuscript in preparation.
- [14] D. Bellus, D.R. Kearns, K. Schaffner, *Helv. Chim. Acta* **1969**, *52*, 971.